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03/10/2006 01:26 PM

TO NCIC HPV@EPA

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Subject Fw: Cyclohexanone Oxime - 1st Versions - Robust

Summary & Test Plan



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Subject Cyclohexanone Oxime - 1st Versions - Robust Summary

& Test Plan

March 10, 2006

To:

Administrator

U.S. Environmental Protection Agency

P.O. Box 1473

Merrifield, VA 22116

Attention: Chemical Right-to-Know Program

From:

Henry J. Trochimowicz, ScD, DABT

Delaware Toxicology Associates, Inc.

Subject: Cyclohexanone Oxime (CAS No. 100-64-1):

1st Versions of a Robust Summary/Test Plan for the HPV Challenge

Program, AR-201

On behalf of <u>DSM Chemicals North America</u>, <u>Inc.</u>, I am submitting the first versions of an HPV Robust Summary and Test Plan for Cyclohexanone Oxime. The preceding two documents are attached as WORD XP files.

Please address any questions or comments concerning this submission to me at:

Delaware Toxicology Associates, Inc. 10 Briarcreek Court, Newark 19711 EMAIL: hitroch@comcast.net

Sincerely yours,

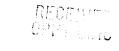
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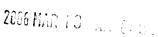


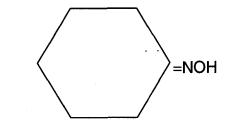
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Cyclohexanone Oxime - SIDS Dossier for EPA.DOC







### CYCLOHEXANONE OXIME

**CAS NUMBER 100-64-1** 

### USEPA HPV CHALLENGE PROGRAM SUBMISSION (FIRST DRAFT)

March 1, 2006

### Submitted by

DSM Chemicals North America, Inc.

### Prepared by:

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### **TEST PLAN**

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### **EXECUTIVE OVERVIEW**

Cyclohexanone oxime, a white crystalline solid, is used primarily as a captive intermediate in the synthesis of caprolactam which, in turn, is polymerized to polycaprolactam (Nylon-6) fibers, resins and plastics. Recent annual production figures for cyclohexanone oxime are not available.

Based on the fact that cyclohexanone oxime is a "closed-system intermediate," and because occupational exposure and releases to the environment are minimal, DSM Chemicals will be providing detailed information in this HPV Test Plan in support of a claim for "reduced testing requirements" for this oxime. This information can be found in an APPENDIX to this Test Plan (See pp. 18-30) entitled "Substantiation of Closed System Intermediate Status."

Adequate data for cyclohexanone oxime are available relative to Physical/Chemical properties. This oxime will be a solid below its melting point (190-196°F) and a liquid above this point. Based on its low vapor pressure (0.029 mm Hg), high boiling point (406°F), and aqueous solubility (1.5 wt%), it will tend to remain in water and only slowly volatilize.

Relative to Environmental Fate and Pathways, limited data are available. Cyclohexanone oxime is stable in water and will hydrolyze only at sustained temperatures (250-300°F). Some data exists on photo-oxidation but there is no information on biodegradation or transport and distribution between environmental compartments. Relative to the category of Ecotoxity, valid date exists for the fathead minnow (96-hr LC50=208 mg/L) but no toxicity information was found for invertebrates and algae. Although most of the preceding studies are not adequate to meet SIDS/HPV requirements, no testing is recommended for the categories of "Environmental Fate and Pathways" and "Ecotoxicity" because cyclohexanone oxime is a "closed system intermediate" and poses no major hazard relative to releases into the environment.

Acute toxicity to mammals appears to be low-to-moderate as demonstrated by an oral lethal dose (LD) in rats of >500mg/kg and a dermal

absorption LD50 in rabbits of >5000 mg/kg. On a repeated exposure basis, several subacute (2-week) and 90-day oral toxicity studies have been conducted in both rats and mice. In the preceding studies, the major target organs appear to be the erythrocyte, the spleen, the bone marrow and liver. Toxicokinetic studies by various routes of administration in rats suggest that cyclohexanone oxime is readily absorbed, subsequently metabolized, and then is excreted in the urine as glucuronides within a day. Relative to genetic toxicity potential, cyclohexanone oxime has been thoroughly tested in both *in vitro* and *in vivo* studies. The overall weight of evidence suggests that cyclohexanone oxime poses no genotoxic hazard. Relative to the HPV Program, adequate studies are available in the areas of "Acute Toxicity", "Repeated Dose Toxicity", and "Genetic Toxicity" and no additional testing is needed.

No definitive studies to assess the potential effects of cyclohexanone oxime on pregnancy or on the reproductive performance of male and female animals have been conducted. However, a determination by EPA that cyclohexanone oxime is a "closed system intermediate" with low occupational exposure potential will eliminate the need for any additional reproductive toxicity testing. A developmental toxicity study, on the other hand, will have to be conducted to fulfill HPV requirements for the "Reproductive/Developmental Toxicity" category. Such a study will be conducted in rats by the oral route and appropriate OECD guidelines will be followed.

Overall, cyclohexanone oxime as a "closed system intermediate" chemical does not appear to represent an unacceptable risk to human health or the environment. Under the EPA HPV Challenge Program, cyclohexanone oxime was evaluated, data gaps were identified, and a decision was made to conduct additional testing only in the area of "Developmental Toxicity". An appropriate study to meet the HPV requirement will reference OECD Guidelines and will be conducted starting in the 3<sup>rd</sup> or 4<sup>th</sup> quarter of 2006 and take less than a year to complete.

Cyclohexanone Oxime

**HPV Test Plan** 

**TESTING PLAN AND RATIONALE** 

### Testing Plan in Tabular Format

Cyclohexanol Oxime	Information Available?	OECD Study?	GLP Study?	Other Study?	Estimation Method?	Acceptable?	Testing Recommended?	Comments
HPV Endpoint								
Physical/Chemical Properties								
Melting Point	Υ	N	N	N	N	Υ	N	
Boiling Point	Y	N	N	N	N	Y	N	
Vapor Pressure	Y	N	N	N	N	Υ	N	
Partition Coefficient	Υ	N	N	N	Υ	Y	N	
Water Solubility	Υ	N	N	N	N	Υ	N	
Environmental Fate								
Photodegradation	Υ	N	N	N	Υ	N	N	*
Water Stability	Υ	N	N	Z		N	N	*
Transport	N						N	*
Biodegradation	N						N	*
Ecotoxicity								
96-Hour Fish	Y	N	N	N	N	Υ	N	*
48-Hour Invertebrate	N						N	*
72-Hour Algae	N						N	*
MammalianToxicity								
Acute Toxicity	Υ	Y/N	Y/N	Υ	N	Υ	N	
Repeated Dose	Υ	Y?	Υ		Ν	Υ	N	
Genotoxicity (Point Mutation)	Υ	Y?	Υ	N	N	Υ	N	
Genotoxicity (Chromosome Aberration)	Υ	Y	Υ	Ν	Z	Υ	N	
Reproductive Toxicity	Y	N	N		N	N	N	*
Developmental Toxicity	N						Υ	Oral rat; OECD Protocol

<sup>\*</sup>Based on claim of "closed system intermediate" status of cyclohexanone oxime and very low potential for both occupational exposure and environmental releases. See attached APPENDIX (Starting on p. 18).

### INTRODUCTION

Cyclohexanone oxime, CAS No. 100-64-1, is a chemical intermediate used primarily in a closed system in the production of caprolactam. The latter chemical is subsequently polymerized to produce Nylon-6 (polycaprolactam) fibers, resins, and plastics.

As part of this HPV Test Plan, DSM Chemicals North America, a primary producer and the HPV Sponsor of cyclohexanone oxime, has provided detailed information in support of a claim for reduced testing requirements for this "closed system intermediate". This information is contained in an APPENDIX to this Test Plan (See pp. 18-30) entitled: "Substantiation of Closed System Intermediate Status." Acceptance of such a status will result in a reduced SIDS testing plan for cyclohexanone oxime.

Various studies have already been conducted on the toxicity of cyclohexanone oxime. Those studies (key and other supporting studies) are summarized in this document with comments as to whether or not they meet the requirements of the USEPA High Production Volume (HPV) Program. Robust summaries, using a SIDS format, have been prepared and include detailed information on key studies and some supporting studies; these detailed summaries are contained in a separate document (Tier 1 Screening SIDS DOSSIER on the HPV Phase....Chemical).

### PHYSICAL-CHEMICAL DATA

Physical/chemical properties for cyclohexanone oxime are available from the literature and from the manufacturer:

Melting Point 190-196°F (1)

Boiling Point 406°F (1)

Vapor Pressure 0.029 mm Hg @ 77°F(1)

Partition Coefficient Log  $P_{ow} = 0.84 @ 77^{\circ}F(2)$ 

Water Solubility 1.5 wt% @ 68°F(1)

Cyclohexanone oxime (MW=113.18) is a 6-carbon ring with an "NOH" group on C1. It is characterized as a white solid at room temperature and as a clear-to-white crystalline liquid above its melting point of 190-196° F(1). It has a specific gravity (water=1) of 0.97 and a pungent-to-slightly sweet odor (1). Cyclohexanone oxime also has a calculated Henry's Law Constant of 8.05E-06 atm-m³/mole (@ 25°C)(2). It also has a lower flammability limit of 1.3%, a flash point (closed cup) of 181.4°F and autoflammability temperature of 545° F (1).

### **Recommendation:**

No additional studies are recommended to fulfill the HPV required end points for "Physical/Chemical Properties".

### **ENVIRONMENTAL FATE AND PATHWAYS**

Atmospheric photo-oxidation may be an important removal process for cyclohexanone oxime. It has a calculated atmospheric OH constant of 7.07E-12 cm³/molecule-sec (2). Relative to stability in water, a manufacturer's MSDS states that the chemical is stable and that hydrolysis occurs only at sustained temperatures (250-300°F)(1). No information was available on Transport and Distribution between Environmental Compartments and no information was available on biodegradation.

### Recommendation:

Although none of the proceeding studies are adequate to meet SIDS/HPV requirements, no additional testing is recommended since cyclohexanone oxime is a "closed system intermediate" and poses no major concerns relative to releases into the environment.

### **ECOTOXICITY**

Acute aquatic toxicity data are available for cyclohexanone oxime in fish. In a study following flo-through guidelines, the 96-hr LC50 based on survival for the fathead minnow (Pimephales promelas) was 208 mg/L (3). No information was available on invertebrates or algae.

### Recommendation:

Although the preceding fish toxicity data may meet HPV requirements, there is no information on invertebrates or algae. However, no additional testing is recommended since cyclohexanone oxime is a "closed system intermediate" and poses no major concern relative to environmental releases.

### MAMMALIAN TOXICITY

### A. Acute Toxicity

The acute toxicity potential of cyclohexanone oxime has been evaluated by several routes of administration. By the intraperitoneal route, its LD50 in mice was 250 mg/kg (3). By an unspecified route of administration, an LD50 of 710 mg/kg was reported for male mice (4).

When cyclohexanone oxime was given orally to rats, its "LD" was reported as >500 mg/kg (5). This value for oral toxicity is supported by results from a 10-dose subacute oral study at 300 mg/kg showing no mortality in rats (6).

By the dermal route of administration, the dermal absorption LD50 in rabbits was >5000 mg/kg, the highest dose tested. Although rabbits showed no adverse clinical signs or body weight changes, various red blood cell parameters were affected and methemoglobin was elevated at all dose levels (800, 2000 and 5000 mg/kg), suggesting that cyclohexanone oxime may be absorbed through the skin in toxicologically significant amounts (7).

There were no reliable data found on inhalation toxicity potential.

However, based on the "closed-system intermediate" status of cyclohexanone oxime, inhalation exposure of workers does not present a significant hazard.

### Recommendation:

The preceding acute toxicity studies by the oral, dermal and intraperitoneal routes are adequate to fulfill HPV requirements for "Acute Toxicity".

### B. Repeated Dose Toxicity

Several repeated dose toxicity studies on cyclohexanone oxime have been conducted by the oral route by both gavage and drinking water administration.

Two 2-week gavage studies were conducted in rats showed dose-related erythroid hyperphasia in the spleen and bone marrow. In one study (8), Sprague-Dawley rats that received 1, 10, or 1000 mg cyclohexanone oxime per kg body weight for 2 weeks had hematologic differences including lower erythrocyte counts, higher platelet counts, lower hemoglobin concentrations and hematocrit levels, and greater mean red cell hemoglobin and mean red cell volume values than the control values. Bone marrow smears indicated lower myeloid, lymphocyte, and monocyte counts concomitant with elevated erythroid counts. There was also general splenic enlargement with hematopoietic cell proliferation.

In a second study (6), male and female F344 rats that received 10, 25, 75, 150, or 300 mg cyclohexanone oxime per kilogram body weight by gavage for 2 weeks had adverse hematologic changes similar to those of the Sprague-Dawley rats. Observations included a dose-related decrease in erythrocyte counts with concomitant increases in the numbers of circulating nucleated erythrocytes and reticulocytes and reduced hematocrit levels and hemoglobin concentrations. Methemoglobin concentrations, measured at the highest dose, were significantly elevated. The rats were observed for another 2 weeks without compound

administration. By Day 28, hematologic values in females had returned to normal and males displayed only slightly depressed erythrocyte counts and mildly elevated reticulocyte counts. No significant effects on body weights and no clinical signs of toxicity were noted in males or females. Splenomegaly and hepatomegaly were observed in male and female rats on Day 14 and Day 28. The hematology results suggested that the hematotoxic effects of cyclohexanone oxime administration were reversible following cessation of exposure. The authors theorized that cyclohexanone oxime induces oxidative damage to the erythrocyte resulting in hemolytic anemia compensated by increased erythropoiesis.

The results of 13-week oral toxicity studies in rats and mice were similar to those of the two-week oral studies with evidence of splenomegaly and erythroid hyperplasia in the spleen and bone marrow. In an oral gavage study (7), Fischer 344 rats (20/sex/dose) received doses of 0, 0.25, 2.5, and 25 mg cyclohexanone oxime per kilogram body weight five times a week for 13 weeks. All males survived to the end of the study; three of 20 females in the 25 mg/kg group died before the end of the study. Males were observed with clinical signs of toxicity that included persistent red nasal discharge (at 25 mg/kg only), chromodacryorrhea and swollen conjunctiva (at 2.5 and 25 mg/kg), and corneal opacity (at all dose levels). No significant effects on body weight or feed consumption were observed in males or females. Hematologic changes similar to those seen in the 2-week study were noted. Dose-related anisocytosis, poikilocytosis, elevated osmotic red blood cell fragility, and a greater incidence of Howell-Jolly bodies were observed. Splenomegaly was noted at necropsy, and histopathologic examination showed erythroid hyperplasia in the bone marrow and spleen and increased hemosiderin pigment deposition in the spleen. Data from satellite groups terminated at 30 and 60 days showed a NOEL at the lowest dose, but results from the end of the study showed a clear cumulative doseresponse down to the 0.25 mg/kg dose level.

In a second 13-week toxicity study (9), B6C3F1 mice (10/sex/dose) were given drinking water containing 0, 625, 1,250, 2,500, 5,000 or 10,000 ppm

cyclohexanone oxime. Deaths occurred in the 10,000 ppm groups and weight gain was depressed in males and females given 10,000 ppm and in females given 5,000 ppm. There were significant increases in relative spleen weight at exposure levels of 5,000 and 10,000 ppm and significant increases in the relative liver weights of males and females that received 10,000 ppm. Microscopically, hemtopoietic cell proliferation was observed in the spleen of males and females in the 5,000 and 10,000 ppm groups. Centrilobular cell hypertrophy was observed in the liver of males in the 2,500, 5,000, and 10,000 ppm groups and in females in the 5,000 and 10,000 ppm groups. Olfactory epithelial degeneration was observed in all exposed groups. In summary, the major targets of cyclohexanone oxime were the erythrocyte, spleen, liver and nasal epithelium. The NOEL for erythrotoxicity is 2,500 ppm following 13 weeks of exposure. The NOEL for hematopoietic cell proliferation in the spleen is 2,500 ppm. The NOEL for hepatotoxicity is 1,250 ppm for males and 2,500 ppm for females following 13 weeks of exposure. Some nasal olfactory epithelial degeneration was observed at all exposure levels; only at 625 ppm in males was the incidence of this lesion not significantly different from that in the controls.

### **Recommendations:**

The subacute and subchronic oral toxicity data on cyclohexanone oxime are adequate to meet the HPV requirements for "Repeated Dose Toxicity".

### C. Genotoxicity

Negative results were obtained in earlier *in vitro* mutogenicity tests with several strains of *Salmonella tyhimurium*, with and without metabolic activation (10, 11) and with *Escherichia coli* strain WP2 (10). In a later point mutation assay (9), cyclohexanone oxime was mutagenic in *Salmonella typhimurium* TA1535 with hamster S9 activation but negative in the same strain with rat liver S9 and negative without any S9 activation. No evidence of mutagenicity was seen in strains TA97, TA98, or TA100 with or without rat or hamster S9

activation. Under similar experimental conditions (12) the same positive result in strain TA1537 was reproduced using hamster liver S9; similarly, no evidence of mutagenicity was seen in strain TA100 with or without hamster liver S9 activation.

In a non-bacterial, *in vitro* mutagenicity assay (9), cyclohexanone oxime tested negative for induction of chromosome aberrations with S9 activation and equivocal in the absence of rat liver S9. In one other *in vitro* assay (11), this oxime was positive in L5178Y mouse lymphoma cells without metabolic activation; the addition of rat liver S9 eliminated the mutagenic effect.

Relative to *in vivo* mutagenicity, cyclohexanone oxime was negative in an intraperitoneal mouse micronucleus study at doses (3 doses at 24 hour intervals) as high as 1000 mg/kg. In addition, this oxime was also negative in a micronucleus assay conducted on mice that were given the chemical at drinking water doses as high as 10,000 ppm for 90 days (9). In one other in vivo study (13), there was no increase in the frequency of sex-linked recessive lethal mutations in germ cells of male *Drosophila melanogaster* administered cyclohexanone oxime by feeding.

Based on an overall weight-of-evidence approach, cyclohexanone oxime is not mutagen.

### **Recommendation:**

No additional testing is required. The HPV requirement for genetic testing has been fulfilled by the preceding *in vitro* and *in vivo* studies sensitive to both point mutations and chromosome aberrations.

### D. Reproductive Toxicity

No definitive studies to assess reproductive performance of male and female experimental animals have been conducted on cyclohexanone oxime.

In a 90-day drinking water study (9) on cyclohexanone oxime, mice receiving drinking water containing as much as 5,000 ppm were given sperm

motility and vaginal cytology evaluations. There were no differences between treated and control mice. In addition, there were no histophathological effects seen in the reproductive organs of the male or female mice.

No other information on the reproductive toxicity potential of cylohexanone oxime was available.

### **Recommendation:**

Although the preceding information does not meet the HPV requirements for "Reproductive Toxicity", no additional testing is recommended. If EPA accepts DSM North America's request to categorize cyclohexanone oxime as a "closed system intermediate", and the supporting data indicating very low exposure potential for both man and the environment, then reproductive toxicity testing will not be a requirement for this oxime.

### E. Developmental Toxicity

No information on the developmental toxicity potential of cyclohexanone oxime was found in the toxicological literature (published or unpublished).

### **Recommendation:**

Since a "closed-system intermediate" categorization of cyclohexanone oxime does not eliminate the HPV requirement for an adequate developmental toxicity study, DSM Chemicals North America will conduct such a study in rats, by the oral route, using the appropriate OECD guidelines.

### F. Toxicokinetics

A toxicokinetic study (14) of cylochexanone oxime has been conducted in male Fischer 344 rats by three different routes of administration. The chemical was found to be rapidly absorbed and cleared within 24 hours after a single oral administration of 1, 10, or 30 mg/kg of [14 C]-cyclohexanone oxime in aqueous

solution. The majority of the cyclohexanone oxime-derived radioactivity was excreted in the urine. Three urinary metabolites were identified: cyclohexylglucuronide and the monoglucuronides of *cis*- and *trans*-cyclohexane-1,2-diol. Low levels of radioactivity (2%-3% of the dose) were retained in the tissues 24 hours after exposure. After intravenous administration of 1 mg/kg of [14 C]-cyclohexanone oxime, the oxime was rapidly cleared from plasma, with half lives of 1.6 minutes (alpha phase) and 18.2 minutes (beta phase). When cyclohexanone oxime was applied dermally (30 mg/kg), only 4% to 5% of the dose was recovered in the urine, feces, and tissues. The majority of the dose volatilized from the skin surface. However, the absorbed radioactivity was readily distributed and excreted, and its metabolic fate was no different than that observed after oral administration.

After a 14-day gavage study (8), cyclohexanone oxime has also been reported to induce increased microsomal activity (aniline hydroxylase and aminopyrine demethylase) in rats treated at a dose of 100 mg/kg body weight. In addition, cyclohexanone oxime has been reported to inhibit the oxidative metabolism of ethanol in rats and mice, an effect similar to that produced in humans as a result of disulfiram administration (15, 16, 17).

From the preceding animal studies, it is evident that cyclohexanone oxime can be absorbed by three different routes of administration. Most absorbed cyclohexanol is metabolized and is subsequently excreted as glucuronides.

### **CONCLUSIONS**

Under the EPA HPV Challenge Program, adequate data to meet HPV requirements are available for cyclohexanone oxime relative to Physical/Chemical Properties, Acute Toxicity, Repeated Dose Toxicity, and Genotoxicity. Although the data available for Ecotoxicity and Environmental Fate and Pathways are limited, no additional studies in these areas are recommended since cyclohexanone oxime is a "closed system intermediate" and poses no exposure hazard relative to releases into the environment. The latter claim also

negates the need for the Reproductive Toxicity requirement. However, the "closed-system intermediate" status and low exposure potential does not alleviate the need for an adequate developmental toxicity study. Such a study, following OECD guidelines, will be conducted in rats by the oral route on cyclohexanone oxime.

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### **APPENDIX**

### SUBSTANTIATION OF CLOSED SYSTEM INTERMEDIATE STATUS FOR CYCLOHEXANONE OXIME

DSM Chemicals North America, Inc., hereby submits a claim for reduced SIDS testing needs for cyclohexanone oxime, a "closed-system intermediate." To support such a claim for reduced testing, the Company has provided detailed information on number of manufacturing sites, process descriptions, monitoring data, presence in products, and transport (if applicable) in this APPENDIX to the HPV Test Plan.

The format of this appendix consists of responses (along with diagrams and tabular data) to a required list of questions (excerpted from the SIDS manual). Based on these responses reflecting a very low-to-negligible exposure potential to workers and the environment, DSM Chemicals believes that the information requirements supporting an exemption claim for reduced SIDS testing have been satisfied. The information requirements follow on pages 19-30 of this document.

### Information Requirements Supporting Exemption Claims for Reduced SIDS Testing Based on Exposure Considerations

### I. Information on sites

A. Number of sites: There is only one (1) site - DSM Chemicals North America, Inc. (DCNA) in Augusta, GA

- B. Basis for "closed process" conclusion at each site:
  - 1) process description in enough detail to clarify the basis for claiming that the process is closed;

See Attachment 1 (p. 22-23) for a process description of cyclohexanone-oxime. A simplified block flow diagram of the process is provided in Figures 1 (p.24) and 2 (p.25).

2) if available, monitoring data showing no detection in any media, including the limits of detection;

As shown in Figure 1 (p.24), a small portion of cyclohexanone-oxime does come into contact with process water, which is discharged to our wastewater treatment plant (WWTP). Attachment 2 (p.26) is provided to show the cyclohexanone-oxime concentration in the combined feed to our on-site WWTP, including the monthly average and mean detection limit (MDL). Attachment 3 (p.27) shows the analysis for cyclohexanone-oxime in the WWTP effluent (Weir III). The analysis shows cyclohexanone-oxime at non-detectable (ND) levels, and a limit of detection is provided also.

3) if monitoring data are unavailable, a statement that no monitoring has taken place and the basis for believing, in the absence of data, that the chemical has not been released and that exposure does not occur.

Monitoring data for vapor emissions is unavailable. However, based on the low vapor pressure of

cyclohexanone-oxime (approximately 25 mmHg at normal operating temperature), and the fact that the surge vessel containing this product is heat traced and insulated, controlled at a fairly constant level, and equipped with a conservation vent, emissions of cyclohexanone-oxime are expected to be at de minimus levels. Tank emission calculation spreadsheets are provided in Table 1 (pp. 28-30) showing working losses to the atmosphere from each surge vessel below 0.4 lb/day.

C. Data on "presence in distributed product" or, in the absence of data, the basis for believing it is not present at levels above trace concentrations.

Cyclohexanone-oxime is used as an intermediate by DCNA to manufacture caprolactam. Attachment 4 (p.31) shows cyclohexanone-oxime analysis performed on our final product (caprolactam) storage tank year-to-date, including the yearly average, mean detection limit (MDL), and the internal DCNA Lab procedure.

### II. Information on transport

If transport also occurs, then in addition to the above, the following should be provided:

- Mode of transport (e.g. water, truck, rail, pipeline)
- Volume (annual)
- Types of consignments (e.g. bulk or drums)
- Controls during transport and transfer at dispatching and receiving sites (placards, labels, etc.)

### **Not Applicable**

III. Supporting evidence from a data search that the chemical is not present in other end products

To the best of our knowledge, cyclohexanone-oxime is used as an intermediate chemical in the manufacture of caprolactam. The caprolactam manufacturing process at DCNA is similar to that of our competitors, and as such, we are reasonably confident that their final product caprolactam will have similar analytical results showing only trace amounts of cyclohexanone-oxime in the final product as does DCNA (see Attachment #4 on p. 31).

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### **ATTACHMENT 1**

Cyclohexanone-oxime, henceforth referred to as oxime, is an intermediate product formed in the production of caprolactam. Oxime is produced within the 2 HPO sections (Sections 26 & 36) of DCNA by the oximation of hydroxylamine (hyam) and cyclohexanaone (anone). The hyam is produced by the catalytic reduction of nitrate within the hyam reactor. Because hyam is unstable in a pure state, an aqueous solution of phosphoric acid, ammonium phosphate, and ammonium nitrate (referred to as Inorganic Process Liquor, or IPL) is used as its carrier. The anone is produced within the 2 Oxanone sections (Sections 35 & 45) of DCNA by the air oxidation of cyclohexane.

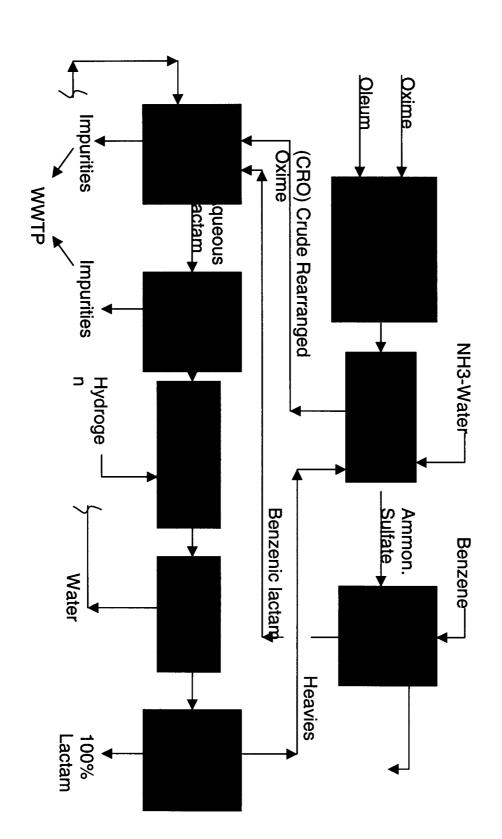
The oximation reaction takes place in 5 mixer-settler reactors where the hyam rich IPL stream is contacted with an organic stream of toluene and anone. The oxime produced in the reaction goes to the organic phase which leaves oximation with an approximate composition of 73% toluene, 25% oxime and 2% anone. This organic stream is washed with water and then distilled within two vacuum distillation columns. The oxime product (see Figure 1), recovered as the bottoms of the section distillation column, is then transferred to rearrangement where it is completely reacted, using oleum as a catalyst, to form caprolactam (see Figure 2). There are 2 rearrangement caprolactam purification sections (Section 27 & 37) at DCNA that further remove impurities and purify the caprolactam to a strength of ~100%.

The only accumulation points for purified oxime within the caprolactam production facility are the pumping vessels between distillation and rearrangement. These vessels, not capable of holding more than 5% of the respective plant's daily production capacity are used to provide just enough surge capacity to enable the safe shutdown of rearrangement or toluene-oxime distillation in the event of a process upset in either of the two sections. During normal operation, the level is controlled at a constant volume in the pumping vessel by making adjustments in the rearrangement section.

Points of release of oxime during the production of caprolactam include wastewater from the HPO sections and some vapor emissions, both of which are minimal. The presence of oxime in the wastewater is primarily the result of the wash step of the toluene oxime and vacuum jet condensate from the toluene/oxime distillation. Prior to discharge, most of the oxime is removed from the washwater via a toluene extraction step. All of the wastewater is routed through a steam stripper, which also removes some oxime. This wastewater is subsequently treated within the site's biological wastewater treatment plant which removes residual oxime to below detectable limits in the plants effluent. The vapor release is limited to that coming from the oxime pump vessel which is limited because the vessel is controlled at a fairly constant level and is equipped with a conservation vent.

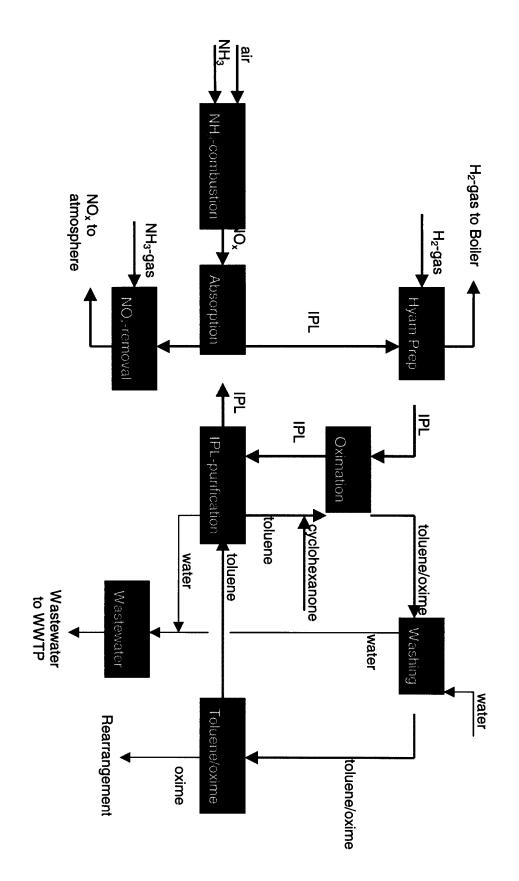
## FIGURE 1

## Rearrangement / Purification (27/37)



## FIGURE 2

# Hydroxylamine Phosphate Cyclohexanone Oxime (26/36)



### **ATTACHMENT 2**

<b>Combined Fee</b>	ed WWTP	
Date	Oxime wt%	
10/31/05	0.0065	
11/01/05	0.0081	
11/02/05	0.0082	
11/03/05	0.0080	
11/04/05	0.0090	
11/05/05	0.0061	
11/06/05	0.0063	
11/07/05	0.0096	
11/08/05	0.0070	
11/09/05	0.0105	
11/10/05	0.0058	
11/11/05	0.0047	
11/12/05	0.0108	
11/13/05	0.0097	
11/14/05	0.0078	
11/15/05	0.0094	
11/16/05	0.0062	
11/17/05	0.0047	
11/18/05	0.0083	
11/19/05	0.0101	
11/20/05	0.0113	
11/21/05	0.0069	Method
11/22/05	0.0068	DCNA-10-GC047
11/23/05	0.0070	
11/24/05	0.0081	
11/25/05	0.0074	
11/26/05	0.0078	
11/27/05	0.0072	
11/28/05	0.0066	
11/29/05	0.0050	
11/30/05	0.0048	
Average	0.0076	
MDL	0.0006	

### **ATTACHMENT 3**

DSM Chemicals North America, Inc.

DSM Laboratory Special Analysis Request Report

DSMLAB: 9823

SUBMITTED: 11/22/2005 REPORTED: 11/29/200

ORIGINATOR: M. Ray

SAMPLE: Weir III

ANALYSIS: Cyclohexanone oxime

PURPOSE:

PRIORITY:

ANALYST: E. Moe APPROVED: Erin R. Moe

DISTRIBUTION: M. Ray, D. Morris, D. Smith, G. Bowen

cyclohexanone oxime; ppm ND

(limit of detection; 6 ppm)

### **TABLE 1 EXPLANATIONS**

From: Pocta, John

Sent: Wednesday, December 28, 2005 8:57 AM

To: Morris, Dean

Subject: Oxime losses from V-2608/V-3608

Dean,

The oxime vapor emissions from oxime pump vessels V-2608/V-3608 are minimal for the following reasons:

- 1. Oxime has a low vapor pressure (approximately 25 mmHg at normal operating temperature),
- 2. The vessels are traced and insulated,
- 3. The vessels are equipped with conservation vents,
- 4. The vessels are controlled at a fairly constant level

Using the subsequent tables on pp. 28 & 29 (Tank Emission Calculation Forms), the estimated oxime emissions are less than 150 lb/yr from each vessel.

John

### **TANK NO. V-2608**

				Ta	ble	<u>:</u> 1					
	<del>                                     </del>	TANK	EMICO				ON FORM				
nui en nui e		IANN	EIVII 33		/AL	CULATI	ON FORI	VI			
ank No.	V-2608		Tank type	Horizonta	al fiv	ed roof (insula	Date		03/04/06		İ
aterial stored	Oxime	-	Company				Performed b	V	00,04,00		
ity	Augusta		State	GA		<del></del>		4			
escription	Outdoor store	age tank									
	INPUT DAT				П		CALCULA	TIONS			1
<del></del>	MEGIDAL	Symbol	<del></del>	Units			CALCOLA	Symbol	<del> </del>	Units	
		Symbol		Office	1		+	Symbol	<del> </del>	OTRUS	<b>!</b>
por pressure Antoir	e constants	<u> </u>			New	EPA method (AF	2-42) *			<del> </del>	
Constant A			9.0490								
Constant B			2,992.500			thing losses			Ĺ		
Constant C		ļ.,	273.150			Tank vapor space	volume	Vv	226.20		
Molecular weight		Mv	113.2	Lb/lb-mole		Vapor density	ingles frietes	KE:	6.531E-03 -0.01433	ID/IT/3	ļ
ınk design data	-	·	<b> </b>			Vapor space expa Vented vapor satu		Ks	0.9063	62	<del> </del>
Shell height		Hs	9.00	ft.		VOIROG Vapor Saco	il dison factor	113	0.5005	102	1
Diameter		D	8.00		Brea	thing losses		LB	-	lb/yr	1
Liquid height			9.00	ft	L						1
Avg. Liquid height		HL	4.50		Work	ing losses		Lw	145.58	lbAyr	4
Tank volume		Į. —		gallons	L			<del> </del>		 	ł
Turnovers	<del> </del>	N Q	120 672	gallons/vr	Tota	losses		LT	145.58	IDAYT	
Net throughput Tunover factor	1	KN	0.899	galionstyf	$\vdash$			<del></del>			ł
Working loss produ	rct factor	Кр	1.00		1	***************************************					• • • • • • • • • • • • • • • • • • • •
eteorological data		† <del></del>	1		Sim	lified method **					1
Daily ave ambient	temp:	TAA	NA	*F							
Daily max. ambient		TAX	N/A	°F		thing losses					1
Daily min. ambient		TAN	N/A	%F 1•F		Temperature expa			#VALUE!	Paras at a data a	<b>i</b>
Daily ambient temp Tank paint solar ab		DTA	NIA NIA	Tr	+	Air displaced per	day		#VALUE!	lbmole/day	<b>.</b>
Daily total insolation		l ox	N/A	Rtuft?_day	Brea	thing losses		LB	#VALUE!	lbAyr	
Darry total illisolation	i lactor	<del>'</del>	TWA-	Distriz-Gay	Dies	ulling losses			WYALOLI	1077	1
Liquid bulk tempera	ature	тв	240,00	•F	Work	ing losses		LW	#VALUE!	lb/yr	
Daily vapor temp. re		DTv	10.00	۴					1		
					Tota	losses		LT	#VALUE!	ib/yr	
Daily ave: liquid sur		TLA	240.00		<b>└</b>						
Daily max liquid su		TLX	242.50 237.50		₩				<del> </del>		<b></b>
Daily min. liquid sur	rrace temp.	1114	237.50	<del></del>	<del> </del>			+	<del> </del>		
VP @ daily ave. liq	uid surf temp	PvA	22.4063	mm Ha	1			1	1		
VP @ daily max. lic		PvX	23.8661					<u> </u>	<u> </u>	<b>†</b>	<u> </u>
VP @ daily min, liq		PvN	21.0264	mm Hg							
VE 6 1	1		h.v.4		$\vdash$				-	1	<b>.</b>
VP @ daily ave an	noient temp.	Pamb	N/A	mm Hġ	1			+	<del>                                     </del>	<del>                                     </del>	<b></b>
Daily vapor pressu	: re range	DPv	2.84	mm Hg	1			+	<del>                                     </del>	<del> </del>	1
Breather vent pressu		15	0.46		1			1	1		1
Breather vent press		DPB		mm Hg	1		i				1
					$\Box$					I	
			ļ	ļ	1			<u> </u>	ļ	ļ	ļ
New EPA method	(Source AD 40	Quantament	E - Ontober	1002)		Special Cases:				ļ	ļ
HAM FLY INACION	TOURISM NE 42 -	Anhiaman		(494)	1		nderground Tanks:	omit breathing	losses (LB)		-
Simplified method	(Adaptation of th	ne new EPA	nethod)	<u> </u>	+		s: use actual liquid t			emp and rand	
				<u></u>		3. Indoor Tanks:	use actual indoor t	emp and range	reduce solar	insolation to	0.
Note - Cells in pin	k are input cells.	All other cell	s are calculat	ed celis.			nservation vents: us				range.
	<u> </u>		<u> </u>			5. Tanks with N2	pads: use AP-42 i	method and en	ter breather ve	ent range.	ļ
	Solar Absorptano			ļ	1 1						
Paint Color	Paint Shade		Factors Condition			Breether Vent D	ange = Pressure v	ant calling	Lau Hiller news	eettina	<u> </u>
		Good	Poor	<b> </b>	-	Disented Astr K	= 10"wc - (-0.5o		TACUUM YEM	ə Turiğ	
Aluminum	Specular	0.39	0.49		1-1		= (16)(.036)psi -		+ 03 psi = 61	DSi	<u> </u>
Aluminum	Diffuse	0.60	0.68		11		Lion.ocomai.	1 .00p.dry00	100 - 1011	T	<u>†</u>
Gray	Light	0.54	0.63	<u> </u>						1	1
Gray	Medium	0.68	0.74								ļ
Red	Primer	0.89	0.91	<b> </b>					<u> </u>	ļ	ļ
White	NA	0.17	0.34	ļ	-						
Black	NA	1.00	1:00	<u></u>						I	L

### **TANK NO. V-3608**

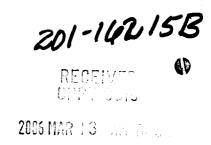
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		TANK	ENNICE	·	ALCULATION	ON EODI	l:A	<u> </u>		
		IANN	EIMI 23		ALCULATI	ON FORI	Áī			
ank No. V	-3608		Tank type	Horizont	al fixed roof (insula	Date		03/04/06		· · · · · · · · · · · · · · · · · · ·
	xime		Company			Performed b	у			
ity A	ugusta		State	GA						
escription 0	utdoor store	ge tank								
IN	PUT DATA	4				CALCULA	TIONS			
The state of the s		Symbol		Units			Symbol		Units	
apor pressure Antoine	constants				New EPA method (AP	-42) *				
Constant A			9.0490							l
Constant B		ļ	2,992.500	ļ	Breathing losses	-1	126.	100.50	62	
Constant C Molecular weight		Mv	273.150	Lb/lb-mole	Tank vapor space v Vapor density	volume	W	188.50 6.531E-03		
Molecular Mergin		IAIA	113.2	COND-INDIA	Vapor space expa	nsion factor	KE	-0.01433	KARIS	
ank design data		<b>†</b>			Vented vapor satur		Ks	0.9207	ft2	
Shell height		Hs	7.50	ft						
Diameter		D	8.00		Breathing losses		LB	-	lb/yr	
Liquid height		ļ	7.50			<u> </u>		140.51		ļ
Avg. Liquid height		HL	3.75		Working losses		Lw	140.81	lb/yr	ļ
Tank volume Turnovers		N	2,820 <b>76</b>	gallons	Total losses		LT	140.81	lbAr.	}
Net throughout		Q		gallons/yr	1048103362	<del>-  </del>	12.	140.61	клуг	
Tunover factor		KN	0,559	ganonaryi			_			<u> </u>
Working loss product f	actor	Кр	1.00							
eteorological data					Simplified method **					
Daily ave ambient ten		TAA	NA	°F						
Daily max, ambient ter		TAX	N/A	°F	Breathing losses	<u> </u>				
Daily min, ambient ten		TAN DTA	N/A	*F	Temperature expan Air displaced per o		+	#VALUEI	ibmole/day	
Daily ambient temp. ra Tank paint solar absor		a	NA	r	All displaced per c	Jey	+	#VALUE!	ichicierday	
Daily total insolation fa		ı .	NA	Btu/ft2-day	Breathing losses		LB	#VALUE!	lb/yr	
		İ	1							
Liquid bulk temperatur	•	TB	240.00		Working losses		Lw	#VALUE1	lb/yr	
Daily vapor temp, rang	je .	DTv	10.00	°F			1			
	<del></del>		240.00	**	Total losses	<u> </u>	LT	#VALUE!	lb/yr	
Daily ave. liquid surfac		TLA	240.00			<del></del>		<del> </del>		
Daily max, liquid surface Daily min, liquid surface		TLX TIN	242.50 237.50			1	+		<del> </del> -	
Daily Hairi. IIquia Sariac	o tomp.	1	207.00	·			1	1	<u> </u>	
VP @ daily ave. liquid	surf. temp.	PvA	22.4063	mm Hg						
VP @ daily max. liquid	surf. temp.	PvX	23.8661							
VP @ daily min. liquid	surf. temp.	PVN	21.0264	mm Hg	<u> </u>				ļ	
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VP @ daily ave. ambi	ык төпкр.	Pamb	N/A	mm Hg				<del> </del>	-	
Daily vapor pressure r	ange	DPv	2.84	mm Hg	1		+			
Breather vent pressure		1	0.46				<u> </u>		1	1
Breather vent pressure		DPB.		mm Hg						I
		ļ							ļ	ļ
		<u> </u>	<u> </u>	ļ		<del> </del>			ļ	Į .
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. 100 LI A MEDICO (C	.v	-apprentel R				derground Tanks:	omit breathing	losses (LB)	•	
" Simplified method (A	daptation of th	e new EPA	method)	<u> </u>		use actual liquid			emp and rand	) <del>0</del> .
					3. Indoor Tanks: (	use actual indoor t	temp and range	; reduce solar	insolation to	0.
Note - Cells in pink a	re input cells. /	All other cell	s are calculat	ed cells.		nservation vents: u				range.
	I 4	1	<u> </u>		5. Tanks with N2	pads: use AP-42	method and en	ter breather ve	ent range.	į
	ler Absorptand aint Shade							1		<del></del>
Panit Color P	ans Shade		Factors Condition		Breather Vent Ra	nge & Preserve	uent setting -	raciulm vens	eelling	<u></u>
		Good	Poor		DISSURE VEIN NE	= 10"wc - (-0.50)			earru ið	1
Aluminum S	pecular	0.39	0.49	<b>†</b>		= (16)(.036)psi -		+.03 psi = .61i	OSi	
Aluminum D	iffuse	0.60	0.68							
I a	ght	0.54	0.63				<u> </u>			
				i .	A CONTRACTOR OF THE CONTRACTOR	:			1	1
Gray M	ledium	0.68	0.74	ļ	ļ			+	ļ	7
Gray M	rimer	0.68 0.89 0.17	0.74 0.91 0.34							

### **ATTACHMENT 4**

		ATTACHMENT 4
T-2801		
Date	Oxime ppm	
01/04/05	0.5	
01/11/05	0.4	
01/18/05	0.6	
01/25/05	0.3	
02/01/05	0.3	
02/08/05	0.2	
02/15/05	0.2	
02/22/05	0.2	
03/01/05	0.3	
03/08/05	0.3	
03/15/05	0.3	
03/22/05	0.3	
03/29/05	0.5	
04/05/05	0.1	
04/12/05	0.3	
04/19/05	0.2	
04/26/05		
05/03/05		
05/05/05		
05/17/05		
05/24/05		
05/24/05		
06/07/05		
06/14/05		
06/21/05		
06/28/05		
07/05/05		
07/12/05		
07/19/05		
07/26/05		
08/02/05		
08/09/05		
08/16/05		
08/23/05		
08/30/05		
09/06/05		
09/13/05		
09/20/05		
09/27/05		
10/04/05		Method
10/11/05		DCNA-10-CP009
10/18/05		
10/25/05		
11/01/05		
11/08/05		
11/15/05		
11/22/05		
11/29/05		
Average		
MDL	0.17	
	Oxime in	

lactam

W.



### TIER 1 SCREENING SIDS DOSSIER ON THE HPV PHASE ..... CHEMICAL

### CYCLOHEXANONE OXIME CAS No. 100-64-1

First Draft March 1, 2006

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	C. OECD NAME	
	D. CAS DESCRIPTOR	_
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# SIDS PROFILE

DATE: November 30, 2005

1.01 A.	CAS No.	108-93-0				
1.01 C.	CHEMICAL NAME	CYCLOHEXANONE OXIME				
1.01 D.	CAS DESCRIPTOR	Not applicable				
1.01 G.	FORMULA & STRUCTURE	C <sub>6</sub> H <sub>12</sub> 0				
1.5	QUANTITY	No current production information available				
1.7	USE PATTERN	Primarily used in a closed process system in the synthesis of caprolactam which, in turn, is used to produce polycaprolactam (Nylon-6) fibers and resins.				
1.9	SOURCES AND LEVELS OF EXPOSURE	Process leaks during manufacture of caprolactam				
TEST PLAN JUSTIFICATION /ISSUES FOR DISCUSSION	No additional testing was recommended for "Environmental Fate and Pathways", "Ecotoxicity", and "Reproductive Toxicity" categories based on "closed system intermediate" status for cyclohexanone oxime (and low occupational and environmental exposure potential) (See APPENDIX (pp. 18-30)in HPV Test Plan document. Adequate studies were available to meet HPV requirements for "Physical/Chemical Properties", "Acute Toxicity" and "Repeated Dose Toxicity" categories. However, a developmental toxicity study in rats by the oral route will be conducted to satisfy HPV requirements for the "Reproductive/Developmental Toxicity" category.					

Tier 1

# SIDS SUMMARY

						March 1		
		Information	OECD Study	GLP	Other Study	Estimation Method	Acceptable	SIDS Testing Required
STUDY		Y/N	Y/N	Y/N	Y/N	Y/N	Y/N	Y/N
PHYSIC	AL-CHEMICAL DATA							
2.1 2.2 2.3 2.4 2.5 2.6	Melting Point Boiling Point Density Vapour Pressure Partition Coefficient a. Water Solubility b. pH and pKa values	Y Y Y Y Y	N N N N N	N N N N			Y Y Y Y Y	и и и и и
2.7 2.8 2.12 2.13	Flash Point Flammability Oxidation: Reduction Potential Adsorption/Desorption to Soil	Y Y N N	N N	N N			Y Y	N N N
ENVIRO	NMENTAL FATE and PATHWAY							
3.1.1 3.1.2 3.3 3.5	Photodegradation Stability in water Transport and Distribution Biodegradation	Y Y N N	N N	N N		Y N	N N	N* N* N* N*
ECOTO	XICITY			]				
4.1 4.2 4.3	Acute toxicity to Fish Acute toxicity to Daphnia Toxicity to Algae <sup>1</sup>	Y N N	N	N			Y	N* N* N*
TOXICI	IY							
5.1 5.1.1 5.1.2 5.1.3	Acute Toxicity: Acute Oral Acute Inhalation Acute Dermal	Y N Y	N N	N N Y			Y Y	N N N
5.1.4 5.4 5.5	Acute intraperitoneal Repeated Dose (General) Genetic Toxicity in vitro	Y	N N	N Y			Y Y	N N
5.6 5.7	Gene mutation     Chromosomal aberration     Genetic Toxicity in vivo     Reproduction Toxicity	Y Y Y N	N N N	Y Y Y			Y Y Y	N N N N*
5.8	Developmental Toxicity/Teratogenicity	N						Y

<sup>\*</sup>Decision based on a claim for "closed system intermediate" status for cyclohexanol oxime based on low occupational exposure potential and negligible environmental release potential; the result of such a status is reduced SIDS testing for this oxime (See APPENDIX (pp. 18-30) of HPV Test Plan document).

#### **GENERAL INFORMATION** 1.

#### SUBSTANCE INFORMATION 1.01

A. CAS-Number

100-64-1

C. OECD Name

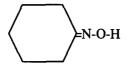
Cyclohexanone oxime

D. CAS Descriptor

Not applicable

G. Structural Formula

C<sub>6</sub>H<sub>11</sub>NO



#### **QUANTITY** 1.5

Remarks:

Cyclohexanone oxime is primarily consumed in a closed system during the production

of caprolactam.

#### 1.7 **USE PATTERN**

Remarks: Most of the cyclohexanone oxime produced is used in the production of caprolactam

during the manufacture of Nylon-6 polymer.

#### SOURCES OF EXPOSURE 1.9

Process leaks during manufacture of caprolactam are remotely possible. However, engineering controls and recommended protective equipment/clothing will assure low exposure potential via inhalation, dermal and eye routes of administration.

#### PHYSICAL-CHEMICAL DATA 2.

#### **MELTING POINT** 2.1

Value:

190 - 196°F

Decomposition:

No Data

Sublimation:

No Data

Method: No Data

GLP: Yes [] No [] ? [X]

Remarks: None

Reliability: [4] Not assignable because limited study information was available.

Reference: DSM Chemicals North America, Inc. Material Safety Data Sheet:

Cyclohexanone Oxime. Julu 31, 1996.

# 2.2 BOILING POINT

Value: 406°F

Pressure: Not available

Decomposition: No Data

Method: No Data

GLP: Yes [] No [] ? [X]

Remarks: No additional data

Reliability: [4] Not assignable because limited study information was available.

Reference: DSM Chemicals North America, Inc. Material Safety Data Sheet:

Cyclohexanone Oxime. Julu 31, 1996.

# 2.3 DENSITY

Type: Bulk density []; Density []; Relative Density [x]

Value: 0.97

Temperature: Not given

Method: No Data

GLP: Yes [] No [] ? [X]

Remarks: No additional data

Reference: DSM Chemicals North America, Inc. Material Safety Data Sheet:

Cyclohexanone Oxime. Julu 31, 1996.

# 2.4 VAPOR PRESSURE

Value:

0.029 mmHg

Temperature:

77°F

Method:

calculated [ ]; measured [ ]

GLP:

Yes [] No [] ? [X]

Remarks:

No additional data

Reliability:

[4] Not assignable because limited study information was available.

Reference:

DSM Chemicals North America, Inc. Material Safety Data Sheet:

Cyclohexanone Oxime. Julu 31, 1996.

# 2.5 PARTITION COEFFICIENT log, P.

log, Pow:

0.84

Temperature:

25°C

Method:

calculated []; measured [X]

Result:

Cyclohexanone oxime log Pow = 0.84

Remarks:

No other information available

Test Substance:

Cyclohexanone oxime (? purity)

GLP:

Yes [] No [] ? [x]

Reliability:

[4] Not assignable because limited study information was available.

Reference:

TOXNET Search on Cyclohexanone Oxime. ChemID Plus Advanced

Search: Physical Properties, September 8, 2005.

# 2.6 WATER SOLUBILITY

Value:

1.5 wt%

Temperature:

68°F

Description:

[]Of very high solubility

[]Of high solubility

[ ]Soluble

[X]Slightly soluble

[]Of very low solubility

[]Not soluble

Method:

No information

GLP: Yes [] No [] ? [X]

Remarks: No additional data

Reliability: [4] Not assignable because limited study information was available

Reference: DSM Chemicals North America, Inc. Material Safety Data Sheet:

Cyclohexanone Oxime. Julu 31, 1996.

2.7 FLASH POINT: 181.4°F (SF Closed Cup)

2.8 AUTO FLAMMABILITY: 545°F

2.9 flammability limits: 1fl = 1.3%

- 2.12 OXIDATION: REDUCTION POTENTIAL No information available
- 2.13 ADSORPTION/DESORPTION TO SOIL No information available

# 3. ENVIRONMENTAL FATE AND PATHWAYS

## 3.1 STABILITY – No information available

# 3.1.1 PHOTODEGRADATION

Type: Air [X]; Water [ ]; Soil [ ]; Other [ ]

Rate constant: 7.07E-12 (cm³/molecules-sec)

Method: Calculated (method unknown)

Remarks: No additional information was available.

Reliability: [4] Not assignable because limited study information was available

Reference: TOXNET Search on Cyclohexanone Oxime. ChemID Plus Advanced

Search: Physical Properties, September 8, 2005.

#### 3.1.2 STABILITY IN WATER

Summary: No specific study to measure hydrolysis in water was found.

However, a manufacturer's MSDS states that cyclohexanone oxime is stable in water and undergoes hydrolysis only at

sustained temperatures (250 - 300°F).

Reliability

(Klimisch Code): [4] Not assignable because limited study information was available

Reference: DSM Chemicals North America, Inc. Material Safety Data Sheet:

Cyclohexanone Oxime. July 31, 1996.

# 3.2 TRANSPORT AND DISTRIBUTION BETWEEN ENVIRONMENTAL COMPARTMENTS INCLUDING ESTIMATED ENVIRONMENTAL CONCENTRATIONS AND DISTRIBUTION PATHWAY – No information available

# 3.5 BIODEGRADATION - No information available

# 4. ECOTOXICOLOGICAL DATA

# 4.1 ACUTE TOXICITY TO FISH

### A. Preferred Result

Type of Test: static []; semi-static []; flow-through [X]; other []

Species: Fathead Minnow

Exposure Period: 96 Hours

Results:  $LC_{so} = 208 \text{ mg/L} (189 \text{ mg/L min to } 230 \text{ mg/L max})$ 

Analytical monitoring: Yes [X] No [ ]

Method: No information available

Test substance: Cyclohexanone oxime (purity unknown)

GLP: Yes [ ] No [ X ]

Remarks: Reported as "not acutely toxic"

Reliability: [4] Not assignable because limited study information was available

Reference: Geiger, D.L., et al. Acute Toxicity of Organic Chemicals to Fathead

Minnows. Volume 5. Center for Lake Superior Environmental Studies,

University of Wisconsin - Superior, WI I: 332, 1990.

# **B.** Supporting Data

One other aquatic toxicity reference which has not yet been located is the following: Applegate, V.C. et al. Toxicity of 4346 Chemicals to Larval Lampreys and Fishes. Spec. Sci. Rep. Fish No. 207, Fish Wildlife Service, U.S.D.I., Washington, D.C., 1957.

# 4.2 ACUTE TOXICITY TO AQUATIC INVERTEBRATES - No information available

# 4.3 ACUTE TOXICITY TO AQUATIC PLANTS (e.g. Algae) - No information available

# 5. TOXICITY

# 5.1.1 ACUTE ORAL TOXICITY

# A. Preferred Result

Type of Test:

LD

Species:

Rat (species/strain unknown)

Value:

>500 mg/kg

Method:

Unknown

Test substance:

cyclohexanol (? purity)

GLP:

Yes [] No [] ? [X]

Remarks:

No information - full reference has not yet been located

Reliability:

[4] Not assignable because limited study information was available

Reference:

National Academy of Sciences, NRC Chemical-Biological Coordination

Center Review 5: 26, 1953

# B. Supporting Data:

Type:

LD

Species:

Fischer 344 Rats (5/sex/dose)

Value:

> 300 mg/kg

Method:

Subacute oral gavage study (10 doses at 300 mg/kg bw)

Test substance:

Purity (>99.5%)

GLP:

Yes [X] No [] ? []

Remarks:

No compound-related mortality after 10 doses at ≤300 mg/kg

Reliability:

[2] Valid with restrictions

A dose of 300 mg/kg was the highest dose tested in this subacute study.

Reference:

Derelanko, M.J., et al. Toxicity of Cyclohexanone Oxime: Hemotoxicity

Following Subacute Exposure in Rats. Fundam, Appl. Toxicol. 5:

117 - 127, 1985.

## 5.1.2 ACUTE INHALATION TOXICITY – No reliable information

# 5.1.3 ACUTE DERMAL TOXICITY

Type: LD50; dermal absorption toxicity

Species: New Zealand albino rabbits (5/sex/dose)

Value: > 5000 mg/kg

Method: Cyclohexanone oxime was applied to the shaved backs of rabbits for 24

hours at dose levels of 0 (distilled water), 0.8, 2 or 5 g/kg under an occluded patch and then observed for 14 days after dosing. Clinical signs and body weights were recorded. Blood samples were taken on days 1, 4 and 7 post-dosing and various hematological and chemical parameters were measured. Animals were terminated after 14 days, spleen weights

were taken, and all rabbits were given gross autopsies.

Test substance: cyclohexanone oxime (99.5% purity)

GLP: Yes [X] No [] ? []

Remarks: No rabbits died at any dose level during the 24-hour dosing period or the

14-day post-dosing period. There were no adverse clinical signs, body weight or organ weight changes associated with treatment. However, reticulocyte counts were elevated on Day 1 in a dose-related manner in males; a similar but not statistically significant elevation occurred in females. Hemoglobin values were depressed in a dose-related manner in females; the depression was statistically significant only at the highest dose at 7 days after dosing. Methemoglobin levels were increased in both sexes in a dose-related manner at 4 days post-dosing, but not at either 1 or 7 days post-dosing. These results suggest that cyclohexanone oxime may be absorbed through the skin in toxicologically significant amounts.

Reliability: [2] valid with restrictions

No mortality occurred at the highest dose tested – 5000 mg/kg, an exceptionally high dose for an acute dermal absorption study.

Reference: Gad, S.C., Derelanko, M.J., Powers, W.J., Mulder, S., Gavigan, F. and

P.C.Babich. Toxicity of Cyclohexanone Oxime: Acute Dermal and Subchronic Oral Studies. <u>Fundam. Appl. Toxicol. 5:</u> 128-136, 1985.

# 5.1.4 ACUTE INTRAPERITONEAL TOXICITY

Type of Test: LD50

Species: Male mice (strain unknown)

Value: 250 mg/kg

Method: No information available

Test Substance: Cyclohexanone Oxime (unknown purity)

GLP: Yes [ ] No [ X ] ? [ ]

Remarks: Limited information available; have not yet located full reference.

Reliability: [4] Not assignable because limited study information was available

Reference: Plzak, V. and J. Doull. National Technical information Services,

No. AD-691490. US Department of Commerce, Washington, D.C., 1969.

## 5.4 REPEATED DOSE TOXICITY

## A. Preferred Result

Type: A 90-Day Oral Gavage Study in Rats

Species/strain: Fischer 344 Male and Female rats (15/sex/exposure level)

Route of Administration: Oral Gavage

Exposure period: 10 rats/sex/dose for 30 days

10 rats/sex/dose for 60 days 20 rats/sex/dose for 90 days

Frequency of treatment: 5 days/week

Post-dosing

observation period: None

Dose Levels: 0, 0.25, 2.5 and 25 mg/kg bw

Control group: Yes (distilled water)

Method: Groups of rats were dosed by oral gavage with cyclohexanol oxime for

5 days/week for 30 days (10 rats/sex/dose), 60 days (10

rats/sex/dose) or 90 days (20 rats/sex/dose) at doses of 0, 0.25, 2.5 or 25 mg/kg body weight. All rats were observed for adverse clinical signs daily and for neurobehavioral effects, body weight changes, and

food consumption on a weekly basis. At dosing termination, hematology, blood chemistry and urinalysis measurements were conducted, as well as a complete histopathological examination of

tissues.

Test Substance: Cyclohexanone Oxime (>99.5% purity)

GLP; Yes [X] No [ ] ? [ ]

Results: There were no significant effects of cyclohenanone oxime on either

body weight or food consumption; a slight mortality occurred at the highest dose (3 female rats) which may or may not have been

treatment-related. In males, treatment-related effects occurred during the first 9 weeks of dosing and included red nasal discharge (highest dose only), chromodacryorrhea and swollen conjunctiva (high and mid

doses), and corneal opacity (all doses). These observations gradually subsided and disappeared by the end of the study. In females, there were no adverse clinical signs during the first 2 weeks of dosing. After that time, adverse signs included chromodacryorrhea (high dose) and corneal opacity (high and mid dose), both of which gradually subsided but never completely disappeared by study termination. Relative to haematology, after 90 days of dosing, there was a doserelated decrease in erythrocytes, hemoglobin and hematocrit, accompanied by an increase in circulating reticulocytes and nucleated erythrocytes, suggesting an increased erythropoeisis in the spleen and bone marrow. The latter changes were confirmed by gross autopsy (splenomegaly) and by histopathological examination. Other than histopathology in spleen and bone marrow, no other organs or tissues were adversely affected. Since the major hematological effects were no severe (no evidence of anemia, e.g.), recovery would be expected upon removal from exposure.

Conclusion:

When rats were exposed repeatedly by oral gavage for up to 90 days, the primary effect of cyclohexanone oxime was increased destruction of erythrocytes with a compensatory increase in erythropoeisis without a noticeable anemia. The bone marrow was able to respond in a sufficient manner to keep up with the added needs. These effects were seen at all dose levels. Since these effects after 90 days of dosing were not severe, recovery would be expected.

**Data Quality** 

(Klimisch Code):

[1] Valid without restrictions

Reference:

Gad, S.C., Derelanko, M.J., Powers, W.J., Mulder, S., Gavigan, F. and P.C.Babich. Toxicity of Cyclohexanone Oxime: Acute Dermal and Subchronic Oral Studies. <u>Fundam. Appl. Toxicol.</u> 5: 128-136, 1985.

**B.** Supporting Results

Type;

A 90-Day Drinking Water Study in Mice

Species/Strain:

B6C3F1 Male and Female Mice

Route of Administration:

Oral (via drinking water)

Frequency of Treatment:

Daily

**Dosing Period:** 

90 Days

**Post-Dosing** 

Observation Period:

None

Dose Levels:

0, 625, 1250, 2500, 5000 and 10000 ppm cyclohexanone oxime in the

drinking water

Control Group:

Yes (water alone)

Method:

Mice (10/sex/dose) were given drinking water containing 0, 625, 1250, 2500, 5000 and 10000 ppm cyclohexanone oxime daily for 90 days. Mice were observed twice daily for mortality and adverse

clinical signs. Clinical observations and body weights were recorded weekly and water consumption was recorded twice weekly. Complete gross and histopathological examinations were conducted at study termination. Sperm motility and vaginal cytology evaluations were performed on mice in the 0, 1250, 2500 and 5000 ppm dose groups. Males were evaluated for necropsy body weight and reproductive organ weights, and epididymal spermatozoal data. Females were evaluated for necropsy body weights, estrous cycle length, and the percent of cycle spent in the various stages.

Test Substance:

Cyclohexanone Oxime (>99% purity)

GLP:

Yes [X] No [ ] ? [ ]

Results:

Deaths occurred in the 10000 ppm groups and weight gain was depressed in males and females given 10000 ppm and also in females given 5000 ppm. There were significant increased in the relative spleen weights at both 5000 and 10000 ppm, and in the relative liver weights of male and female mice dosed at 10000 ppm. Microscopically, hematopoeitic cell proliferation was seen in the spleens of males and females in both the 5000 and 10000 ppm groups. In the liver, centrilobular cell hypertrophy was seen in males at 2500, 5000 and 10000 ppm and in females at 5000 and 10000 ppm. Olfactory epithelial degeneration was seen in all dose groups. There were no significant differences in sperm motility of vaginal cytology parameters between dosed and control males and females.

Conclusion:

The major targets of cyclohexanone oxime administered in the drinking water for 90 days to mice were the erythrocyte, spleen, liver and nasal epithelium. The NOEL for erythrotoxicity and hematopoeitic cell proliferation in the spleen was 2500 ppm. The NOEL for hepatotoxicity was 1250 ppm for males and 2500 for females following 13 weeks of dosing. Some nasal olfactory epithelial degeneration was observed at all dose levels; only at 625 ppm in males was the incidence of this lesion not significantly different from controls. There were no effects on sperm motility or vaginal cytology parameters at doses as high as 5000 ppm (highest dose evaluated).

Data Quality (Klimisch Code):

[1] Valid without restrictions

Reference:

Burka, L.T. NTP Technical Report on Toxicity Studies of

Cyclohexanone Oxime. NTP Report Series No. 50, NIH Publication

No. 96-3934, 1996.

## 5.5 GENETIC TOXICITY IN VITRO

# A. Bacterial In Vitro Test

(1) Type:

Bacterial reverse mutation assay

System of testing: Preincubation protocol

Concentration: cyclohexanone oxime concentrations ranged from 33 µg/plate to 3333

μg/plate (with metabolic activation) and from 333 to 6666 μg/plate (without

metabolic activation); at least 5 doses tested

Method of Activation: With []; Without []; With and Without [X]; No data []

Results: Not mutagenic in Salmonella typhimurium strains TA97, TA98, and

TA100, with or without S9 activation. Positive evidence of mutagenicity only in strain TA1535 with hamster S9 activation but negative in same

strain with rat liver S9 and negative without any S9 activation.

Test Substance: Cyclohexanone Oxime (>99% purity)

Cytotoxicity

Concentration: >3333 µg/plate with S9 activation; >6666 µg/plate without S9 activation

Precipitation

Concentration: No data

Method: Testing was performed as reported by Zeiger (Environ. Mol. Mutagen. 19

(Suppl. 21): 2-14, 1992). Cyclohexanone oxime was incubated with Salmonella typhimurium tester strains (TA97, TA98, TA100 and TA1535) either in buffer (without activation) or S9 mix (metabolic activation enzymes and cofactors from Aroclor 1254-induced male Sprague-Dawley

rats or Syrian hamster liver) for 20 minutes at 37°C. Top agar

supplemented with 1-histidine and d-biotin was added and the contents of all tubes were mixed and poured onto the surfaces of minimal glucose agar plates. Histidine-independent mutant colonies arising on these plates were counted following incubation for 2 days at 37°C. Each trial consisted of triplicate plates of concurrent positive and negative controls and of at least 5 doses of cyclohexanone oxime. All positive assays were repeated under the conditions that elicited a positive response; all negative assays were

also repeated.

GLP: Yes [X] No []?[]

Reliability: [2] valid with restrictions

This oxime was positive in TA1535 but not in TA100, a more sensitive

strain for the same kind of mutation.

Reference: Burka, L.T. NTP Technical Report on Toxicity Studies of Cyclohexanone

Oxime. NTP Report Series No. 50, NIH Publication No. 96-3934, 1996.

(2) Type: Other Point Mutation Assays (Supporting Data)

#### Summary:

Under similar experimental conditions, Prival (2001) reproduced the preceding positive result in strain TA1535 using hamster liver S9, without evidence of mutagenicity in strain TA 100.

However, negative results with cyclohexanone oxime were obtained in mutagenicity tests with several strains of *Salmonella typhimurium*, with and without metabolic activation (Araki 1986; Rogers-Back 1988) and with Escherichia coli strain WP2 (Araki 1986). The only other mutagenic activity reported for cyclohexanol oxime was noted in L5178Y mouse lymphoma cells treated in the absence of S9 activation; the addition of rat liver S9 eliminated the mutagenic effect (Rogers-Back 1988).

#### References:

Araki, A., et al. Mutagenicity of Oxime Compounds in the S. typhimurium TA98, TA100, TA2637, and E. coli WP2 uvrA/pKM101. Mutat. Res. 164: 263, 1986.

Prival, M.J. Anomalous mutagenicity profile of cyclohexanone oxime in bacteria: cell survival in background lawns. <u>Mutat. Res.</u> 497: 1-9, 2001.

Rogers-Back, A.M. et al. Genotoxicity of 6 Oxime Compounds with Salmonella-Mammalian-Microsome Assay and Mouse Lymphoma TK Assay. Mutat. Res. 204: 149-162, 1988.

#### B. Non-Bacterial In Vitro Test

Type: Cytogenetic assay (chromosome aberration)

System of testing: Chinese hamster ovary (CHO) cells

Concentration: Doses of cyclohenanone oxime ranging from 500 to 5000  $\mu$ g/ml

Metabolic activation: With []; Without []; With and Without [X]; No data []

Results: Negative

Cytotoxicity Concentration: >>5000 μg/ml

Precipitation Concentration: 5000 µg/ml

Genotoxic effects: None

Method: Testing was performed as reported by Galloway (Environ. Mol.

Mutagen. 10 (Suppl. 10): 1-175, 1987). Cyclohexanone oxime was tested in cultured CHO cells for induction of chromosome aberrations (Abs), both in the presence and absence of Aroclor 1254-induced male Sprague-Dawley rat liver S9 and cofactor mix. Each test consisted of concurrent solvent and positive controls and of at least 3 doses of cyclohexanone oxime. In the absence of toxicity, 5000 µg/ml was selected as the high dose. A single flask per dose was used; tests yielding equivocal or positive results were repeated. In the ABS test without S9, cells were incubated in McCoy's 5A medium with cyclohexanone oxime for 10 hours; Colcemid was added and incubation was continued for 2 hours. The cells were then harvested by mitotic shake-off, fixed, and stained with Giemsa. For the Abs test with S9, cells were treated with cyclohexanone oxime and S9 for 2 hours, the treatment medium was removed, and the cells were then incubated for 10 hours in fresh medium. Colcemid was added for the

final 2 hours. Cells were then harvested in the same manner as for

treatment without S9. For scoring, cells were selected on the basis of good morphology and completeness of karyotype(21±2 chromosomes) and all slides were scored blind. Two hundred first-division

metaphase cells were scored at each dose level.

GLP: Yes [X] No [] ? []

Test substance: Cyclohexanone Oxime (>99% purity)

Remarks: None

Reliability: [2] Valid with restrictions

Reference: Burka, L.T. NTP Technical Report on Toxicity Studies of Cyclohexanone

Oxime. NTP Report Series No. 50, NIH Publication No. 96-3934, 1996.

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# 5.6 GENETIC TOXICITY IN VIVO

(A) Type: Micronucleus Assay

Species/strain: B6C3F1 Mice

Sex: Female []; Male []; Male/Female [X]; No data []

Route of Administration: Oral (drinking water); intraperitoneal injection

Dosing Period: 90 Days; over 3 days at 24-hour intervals ......

Doses: 0, 625, 1250, 2500, 5000, and 10000 ppm in the water;

400, 600, 800 and 1000 mg/kg (ip).

Results: Negative in 5 mice/sex dose (oral study) and in 5 male mice (ip study)

Effect on mitotic

index or P/N ratio: No information

Genotoxic effects: Not an in vivo mutagen

Method: A detailed discussion of this micronucleus assay on peripheral blood has

been presented by MacGregor (<u>Fundam. Appl.Toxicol. 14</u>: 513-522, 1990). At the end of a 90-day drinking water study on cyclohexanone oxime, peripheral blood samples were taken from 5 mice/sex/dose (highest dose in the drinking water was 10000 ppm), smears were immediately prepared and fixed in absolute methanol, and the slides were then stained with a chromatin-specific fluorescent dye and coded. Two thousand normochromatic erythrocytes were scored in each of 5 mice/sex

in each of the 5 dose groups. The criteria of Schmid (In "Chemical Mutagens: Principles and Methods for their Detection", Vol. 4, A.Hollander (Ed.), pp. 31-53, Plenum Press, New York, 1976) were

used in defining micronuclei.

For the intraperitoneal micronucleus test, after preliminary rangefinding, 5 male mice/dose were injected (ip) over 3 days at 24-hour intervals with cyclohexanone oxime dissolved in corn oil (total dose volume of 0.4 ml) at doses of 0, 400, 600, 800 and 1000 mg/kg bw. Solvent control animals received 0.4 ml of corn oil only and positive control mice got injections of cyclophosphamide. Twenty-four hours after the third injection, the mice were sacrificed and smears of the bone marrow cells (from the femur) were prepared. Air-dried smears were fixed and stained; 2000 polychromatic erythrocytes were scored for frequency of micronucleated cells in each of 5 mice at each of 4 doses.

GLP: Yes [X] No [] ? []

Test substance: Cyclohexanone Oxime (>99% purity)

Remarks: In micronucleus tests conducted in mice by two different routes of

administration (oral and ip), cyclohexanone oxime showed no

evidence of in vivo mutagenicity.

Reliability: [1] Valid without restrictions

Reference: Burka, L.T. NTP Technical Report on Toxicity Studies of Cyclohexanone

Oxime. NTP Report Series No. 50, NIH Publication No. 96-3934, 1996

(B) Type: Gene Mutation In Vivo (Supporting Data)

Summary: When male fruit flies (*Drosophila melanogaster*) were administered

cyclohexanone oxime (8.8 mM) by feeding, there was no increase in the

frequency of sex-linked recessive mutations in germ cells.

Reference: Vogel, E. and J.L.R. Chandler. Mutagenicity Testing of Cyclamate and

Some Pesticides in Drosophila Melanogaster. Experientia 30: 621-623,

1974.

# 5.7 TOXICITY TO REPRODUCTION - No data available

#### 5.8 DEVELOPMENTAL TOXICITY: No data available

# 5.11 EXPERIENCE WITH HUMAN EXPOSURE (WORKPLACE)

No definitive studies on human exposure to cyclohexanone oxime were found. No occupational exposure limits(OSHA PEL or ACGIH TLV® have been established. In one older reference(Finkel, A.J. in Hamilton and Hardy's Industrial Toxicology, 4<sup>th</sup> Edition, John Wright PSG, Boston, MA, 1983), hematological disorders were reported in humans exposed to cyclohexanone oxime. It was also stated that dermatitis

and skin sensitization may also be potential effects of occupational exposure. No other details were given.

# 6.0 **REFERENCES**